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## LAMELLAR POST FOAMING CLEANSING COMPOSITION AND DISPENSING SYSTEM

### 5 BACKGROUND OF THE INVENTION

#### FIELD OF THE INVENTION

10 The present invention relates to detergent compositions suitable for topical application for cleansing the human body, such as the skin and hair. In particular, it relates to self-foaming lamellar compositions, and to aerosol barrier dispensing systems to dispense the self-foaming lamellar compositions.

### 15 THE RELATED ART

The following all disclose post-foaming isotropic gels: PCT publication no. WO 9703646 to C. Hall, et al., Published February 6, 1997 discloses a post-foaming isotropic gel with a base material consisting of a  
20 detergent and a thickener, of viscosity of at least 9,500 cps; US Patent No. 4772427 to A. Dawson, et al. issued September 20, 1988, discloses a post-foaming isotropic gel shower product with a viscosity of between 20,000-100,000 cps; EPO publication no. EP 0987018 to T. McGee et al., published March 22, 2000, discloses a post-foaming isotropic shower gel with a viscosity  
25 of between 1000 –60,000 cps; US Patent No. 5248495 to T. Patterson, et al., issued September 28, 1993, discloses a post foaming shaving gel composition having a viscosity of 15-16,000 cps.

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The following all disclose mousses or moisturizing personal cleansing compositions with aerosol propellants: PCT publication no. WO9938491 to T. Dixon, et al. published August 5, 1999 discloses moisturizing personal cleansing compositions with improved lipid deposition with a viscosity of 300-  
5 100,000 cps and a skin lipid deposition of greater than 25 mg/cm<sup>2</sup> on the skin; PCT publication no. WO9827936 to G. Dawson, et al., Published July 2, 1998, discloses a packaged personal cleansing product with a viscosity less than 700 mPa.s using a Brookfield #2 spindle at 60 rpm and NO or CO<sub>2</sub> or mixtures thereof as foaming agents; US Patent No. 5002680 to R. Schmidt, et al.  
10 issued March 26, 1991 discloses a mild skin cleansing aerosol nonsoap mousse with skin feel and moisturization benefits; PCT publication no. WO9938490 to T. Dixon, Published August 5, 1999 discloses an aerosol personal cleansing emulsion compositions which contain low vapor pressure propellants.

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US Patent No. 5186857 to M. VISHNUPAD et al., issued February 16, 1993, discloses a self-foaming high oil containing, non-lamellar composition of at least 10% having at least 5 % of one high foaming surface active agent which increases the solubility of said foam-producing liquid in said oil or  
20 oil/wax mixture; one or more foam suppressing agents which cooperate with said high foaming surface active agents to help solubilize the hydrocarbon in the oil mixture, and a process for making and using same,

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US Patent No. 5091111 to P. Neumiller, issued February 25, 1992 discloses a vesicular structured aqueous emulsion containing nonionic surfactants and an aerosol delivery system. However the composition taught contains no anionic surfactants or emollients.

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1 The above patents and publications however, fail to disclose or  
2 suggest a post-foaming cleansing composition based on a lamellar structured  
3 fluid that is shear thinning and provides excellent stability for high  
4 concentrations of lipophillic emollients, greater moisturization and sensory  
5 benefits, and greater and more consistent dispensing control than prior art  
6 post-foaming cleansing compositions. Furthermore, the above patents and  
7 publications fail to teach a cost effective dispensing method to dispense the  
8 inventive composition using an aerosol barrier can, preferably a piston type  
9 can. The inventive cleansing composition can find application for body  
10 washes, shaving creams, hair mousses, shampoos, conditioners, scalp  
11 treatments, hand or body lotion, and facial creams. Post foaming is herein  
12 defined as a material which remains substantially free from foaming for a short  
13 time after dispensing from a pressurized can and eventually forms a higher  
14 density product as compared to a mousse which is dispensed immediately as  
15 a foam and forms a lower density product. Prior art mousse compositions are  
16 disadvantageous because they tend to break down and dissipate quickly after  
17 being applied to the skin. Post-foaming gels are generally packaged in  
18 aerosol barrier containers which separate the propellant from the blended  
19 cleansing and foaming agent product. Mousses are packaged with the  
20 propellant blended with the cleansing product.

#### SUMMARY OF THE INVENTION:

21 In one aspect the present invention comprises a liquid cleansing and  
22 moisturizing composition comprising: (a) from about 80 to about 97% by wt. of  
23 a neat cleansing lotion having about 0.5 to about 65% by wt. of the total  
24 composition of at least one non-soap anionic or mixture of non-soap anionic  
25 surfactants; about 35 to about 90% by wt. of the total composition of water;

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(b) from about 3 to about 20% by wt. of the total composition of at least one volatile foaming agent, and wherein the neat cleansing lotion is a lamellar structured shear thinning composition at 25 C.

5           In order to convey a heaping lotion appearance and excellent sensory characteristics, the inventive neat cleansing lotion preferably has a shear thinning index greater than about 0.4, more preferably greater than about 0.5 . Shear thinning index is defined below.

10           In another embodiment, a liquid cleansing and moisturizing composition and dispensing system is provided comprising: (a) from about 80 to about 97% by wt. of a neat cleansing lotion having about 0.5 to about 65% by wt. of the total composition of at least one non-soap anionic or mixture of non-soap  
15   anionic surfactants; about 35 to about 90% by wt. of the total composition of water; (b) from about 3 to about 20% by wt. of the total composition of a volatile foaming agent; wherein the neat cleansing lotion is a lamellar structured shear thinning composition at 25 C; and wherein the cleansing  
20   composition is contained in an aerosol pressurized container having a barrier separating the cleansing composition from an aerosol propellant preferably a piston type can.

          In another embodiment, the inventive liquid cleansing and moisturizing composition and dispensing system is substantially soap free and comprises:  
25   a) from about 80 to about 97% by wt. of a neat cleansing lotion having about 0.5 to about 65% by wt. of the total composition of at least one non-soap anionic or mixture of non-soap anionic surfactants; less than about 4%, preferably less than about 1%, and most preferably less than about 0.5% of a

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soap; about 35 to about 90% by wt. of the total composition of water; and (b) from about 3 to about 20% by wt. of the total composition of a volatile foaming agent; and wherein the neat cleansing lotion is a lamellar structured shear thinning composition at 25 C; and wherein the cleansing composition is  
5 contained in an aerosol pressurized piston container having an aerosol propellant, preferably a piston type can.

10 BRIEF DESCRIPTION OF THE DRAWINGS:

The foregoing features, advantages, and objects of this invention are now described in more detail with reference to the drawings in which:

15 FIG. 1 is a graph of the mean visual scores for the dryness data from the skin moisturization study.

FIG. 2 is a graph of the average replicate scores for the Skicon data from the skin moisturization study.

20 FIG. 3 is a graph of the average replicate scores for the Corneometer data from the skin moisturization study.

FIG. 4 is a graph of viscosity versus shear stress for comparative commercially available Edge ® gel at 4°C.

25 FIG. 5 is a graph of viscosity versus shear stress for inventive lamellar lotion concentrates EE and GG with and without added heptane compared to

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comparative isotropic lotion base concentrate A with and without added heptane.

FIG. 6 is an expanded version of FIG 5 showing in more detail  
5 comparative isotropic lotion base concentrate A with and without added heptane.

FIG. 7 shows a side by side comparison of the stability of the foam produced by an inventive lamellar post foaming lotion compared to a  
10 comparative isotropic post-foaming gel.

#### DETAILED DESCRIPTION OF THE INVENTION:

15 In one aspect the present invention comprises a liquid cleansing and moisturizing composition comprising:

(a) from about 80 to about 97% by wt. of a neat cleansing lotion having about 0.5 to about 65%, preferably 1 to about 25% by wt. of the total  
20 composition of at least one non-soap anionic or mixture of non-soap anionic surfactants;

about 35 to about 90% by wt. of the total composition of water;

(b) from about 3 to about 20% by wt. of the total composition of at least  
25 one volatile foaming agent, preferably having at least one or more hydrocarbons or mixture thereof; and

wherein the neat cleansing lotion is a lamellar structured shear thinning composition at 25 C.

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Advantageously, the neat cleansing lotion further comprises about 0.1 to about 25% by wt. of the total composition of a surfactant selected from amphoteric, zwitterionic or mixtures thereof. Preferably the neat cleansing lotion further comprises about 0.5 to about 50%, more preferably 6 to about 35 by wt. of the total composition of lipophilic emollients, humectants, and mixtures thereof. Advantageously the lipophilic emollient is selected from a triglyceride oil, mineral oil, petrolatum, and a blend thereof, and the like; and the humectants are selected from polyhydric alcohols, polyols, and blends thereof, and the like.

Advantageously, the neat cleansing lotion contains at least one lipophilic emollient in a concentration greater than about 10 %, preferably greater than about 12% by weight of the total composition. Preferably this lipophilic emollient includes at least one triglyceride oil.

Advantageously the inventive neat cleansing lotion contains about 0.1% to about 15% by wt. of the total composition of a lamellar phase inducing structurant selected from: C8 to C24 alkenyl or branched alkyl fatty acid or ester thereof with a melting point below 25C; C8 to C24 alkenyl or branched alkyl fatty alcohol or ether thereof with melting point below 25C; C5 to C12 alkyl fatty acids; and hydroxystearin, and the like. Preferably the lamellar structurant is selected from isostearic acid, lauric acid, oleic acid, palm kernel acid, coconut acid, and blends thereof, and the like.

In order to convey a heaping lotion appearance and excellent sensory characteristics, the inventive neat cleansing lotion has a shear thinning index greater than about 0.4, preferably greater than about 0.5 . Shear thinning index is defined below.

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Preferably the inventive neat cleansing lotion further comprises at least one cationic skin conditioning agent, preferably present in the range of from about 0.01 to about 5 %, more preferably about 0.1 to about 1% by wt. of the total composition. Advantageously the cationic skin conditioning agent is selected from cationic polysaccharides, cationic copolymers of saccharides and synthetic cationic monomers, synthetic cationic polymers, polymeric quaternary ammonium salts of hydroxyethylcellulose, cationic proteins, and salts and derivatives thereof, and the like.

With regards to the surfactants present in the inventive neat cleansing lotion, the anionic surfactant is preferably selected from alkyl ether sulfate, alkyl sulfate, acyl isethionate, mono- and di-alkyl phosphate, and blends thereof, and the like, and the amphoteric /zwitterionic surfactant is preferably selected from cocoamidopropyl betaine, sodium lauroamphoacetate, sodium cocoamphoacetate and blends thereof, and the like.

The combined inventive cleansing lotion and foaming agent has sufficient initial viscosity to convey excellent sensory feel. Preferably the initial viscosity is greater than about 40,000 cps measured at 10 Pa at 25 C according to the controlled shear stress rheological method defined below. More preferably the initial viscosity is in the range of about 40,000 to about 2,000,000 cps at 10 Pa and 25°C.

The inventive neat cleansing lotion advantageously comprises a solubilizing agent, preferably selected from isopropyl palmitate and isopropyl myristate, and the like. In addition, the inventive neat cleansing lotion advantageously contains about 0.1% to about 5% by wt. of a lamellar



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stabilizing material consisting of a polymeric hydrophilic emulsifier modified at one or both ends with hydrophobic polyhydroxy fatty acid ester chain; preferably the emulsifier is dipolyhydroxystearate. In addition, the emulsifier preferably has a polyalkylene glycol backbone chain of general formula: H (O (CH<sub>2</sub>)<sup>a</sup> nOH wherein a is 2 to 4 and n is 2 to 60 having 1 to 50 C<sub>8</sub> to C<sub>24</sub> fatty acid group or groups attached to one or both sides of the backbone. More preferably the fatty acid group or groups attached to backbone chain are selected from hydroxystearic acid, palmitic acid, and blends thereof, and the like.

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In another embodiment, a liquid cleansing and moisturizing composition and dispensing system is provided comprising: (a) from about 80 to about 97% by wt. of a neat cleansing lotion having about 0.5 to about 65%, preferably about 1 to about 25 by wt. of the total composition of at least one non-soap anionic or mixture of non-soap anionic surfactants; about 35 to about 90% by wt. of the total composition of water; (b) from about 3 to about 20% by wt. of the total composition of a volatile foaming agent, preferably a hydrocarbon foaming agent; wherein the neat cleansing lotion is a lamellar structured shear thinning composition at 25 C; and wherein the cleansing composition is contained in an aerosol pressurized container having a barrier separating the cleansing composition from an aerosol propellant, more preferably an aerosol piston can.

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Advantageously, the composition dispensed by the inventive dispensing system has a dynamic density of greater than about 0.2 g/ml, preferably about 0.4 g/ml as measured within 30 seconds after dispensing at 25 C and 1 atm pressure using the methodology provided below.

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In another embodiment, the inventive liquid cleansing and moisturizing composition and dispensing system comprises: a) from about 80 to about 97% by wt. of a neat cleansing lotion having about 0.5 to about 65%, preferably about 1 to about 25% by wt. of the total composition of at least one non-soap anionic or mixture of non-soap anionic surfactants; less than about 4%, preferably less than about 1%, and most preferably less than about 0.5% of a soap; about 35 to about 90% by wt. of the total composition of water; and (b) from about 3 to about 20% by wt. of the total composition of a volatile foaming agent preferably containing at least one volatile hydrocarbon; and wherein the neat cleansing lotion is a lamellar structured shear thinning composition at 25 C; and wherein the cleansing composition is contained in an aerosol pressurized piston container having an aerosol propellant, preferably a piston type can.

#### Lamellar Cleansing Lotion Microstructure

The rheological behavior of all surfactant solutions, including liquid cleansing solutions, is strongly dependent on the microstructure, i.e., the shape and concentration of micelles or other self-assembled structures in solution. When there is sufficient surfactant to form micelles (concentrations above the critical micelle concentration or CMC), for example, spherical, cylindrical (rod-like) or discoidal micelles may form. As surfactant concentration increases, ordered liquid crystalline phases such as lamellar phase, hexagonal phase or cubic phase may form. The lamellar phase, for example, consists of alternating surfactant bilayers and water layers. These layers are not generally flat but fold to form submicron spherical onion like structures called vesicles or liposomes. The hexagonal phase, on the other hand, consists of long cylindrical micelles arranged in a hexagonal lattice. In

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general, the microstructure of most personal care products consist of either spherical micelles; rod micelles; or a lamellar dispersion.

As noted above, micelles may be spherical or rod-like. Formulations having spherical micelles tend to have a low viscosity and exhibit newtonian shear behavior (i.e., viscosity stays constant as a function of shear rate; thus, if easy pouring of product is desired, the solution is less viscous and, as a consequence, it doesn't suspend as well). In these systems, the viscosity increases linearly with surfactant concentration.

Rod micellar solutions are more viscous because movement of the longer micelles is restricted. At a critical shear rate, the micelles align and the solution becomes shear thinning. Addition of salts increases the size of the rod micelles thereof increasing zero shear viscosity (i.e., viscosity when sitting in bottle) which helps suspend particles but also increases critical shear rate (point at which product becomes shear thinning; higher critical shear rates means product is more difficult to pour).

Lamellar dispersions differ from both spherical and rod-like micelles because they can have high zero shear viscosity (because of the close packed arrangement of constituent lamellar droplets), yet these solutions are very shear thinning (readily dispense on pouring). That is, the solutions can become thinner than rod micellar solutions at moderate shear rates.

In formulating liquid cleansing compositions, therefore, there is the choice of using rod-micellar solutions (whose zero shear viscosity, e.g., suspending ability, is not very good and/or are not very shear thinning); or lamellar dispersions (with higher zero shear viscosity, e.g. better suspending,

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and yet are very shear thinning). The use of lamellar dispersions is important for the present invention.

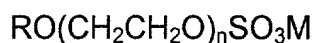
5            Surfactants:

Surfactants are an essential component of the inventive self-foaming cleansing composition. They are compounds that have hydrophobic and hydrophilic portions that act to reduce the surface tension of the aqueous solutions they are dissolved in. Useful surfactants can include anionic, nonionic, amphoteric, and cationic surfactants, and blends thereof.

15            Anionic Surfactants:

The self-foaming cleansing composition of the present invention contains one or more anionic detergents. The anionic detergent active which may be used may be aliphatic sulfonates, such as a primary alkane (e.g., C<sub>8</sub>-C<sub>22</sub>) sulfonate, primary alkane (e.g., C<sub>8</sub>-C<sub>22</sub>) disulfonate, C<sub>8</sub>-C<sub>22</sub> alkene sulfonate, C<sub>8</sub>-C<sub>22</sub> hydroxyalkane sulfonate or alkyl glyceryl ether sulfonate (AGS); or aromatic sulfonates such as alkyl benzene sulfonate.

The anionic may also be an alkyl sulfate (e.g., C<sub>12</sub>-C<sub>18</sub> alkyl sulfate) or alkyl ether sulfate (including alkyl glyceryl ether sulfates). Among the alkyl ether sulfates are those having the formula:



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wherein R is an alkyl or alkenyl having 8 to 18 carbons, preferably 12 to 18 carbons, n has an average value of greater than 1.0, preferably greater than 3; and M is a

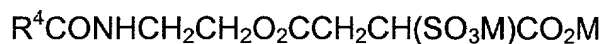
solubilizing cation such as sodium, potassium, ammonium or substituted ammonium. Ammonium and sodium lauryl ether sulfates are preferred.

The anionic may also be alkyl sulfosuccinates (including mono- and dialkyl, e.g., C<sub>6</sub>-C<sub>22</sub> sulfosuccinates); alkyl and acyl taurates, alkyl and acyl sarcosinates, sulfoacetates, C<sub>8</sub>-C<sub>22</sub> alkyl phosphates and phosphates, alkyl phosphate esters and alkoxyl alkyl phosphate esters, acyl lactates, C<sub>8</sub>-C<sub>22</sub> monoalkyl succinates and maleates, sulphoacetates, alkyl glucosides and acyl isethionates, and the like.

Sulfosuccinates may be monoalkyl sulfosuccinates having the formula:

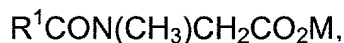


amide-MEA sulfosuccinates of the formula;



wherein R<sup>4</sup> ranges from C<sub>8</sub>-C<sub>22</sub> alkyl and M is a solubilizing cation.

Sarcosinates are generally indicated by the formula:



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wherein R<sup>1</sup> ranges from C<sub>8</sub>-C<sub>20</sub> alkyl and M is a solubilizing cation.

5 Taurates are generally identified by formula:



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wherein R<sup>2</sup> ranges from C<sub>8</sub>-C<sub>20</sub> alkyl, R<sup>3</sup> ranges from C<sub>1</sub>-C<sub>4</sub> alkyl and M is a solubilizing cation.

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The inventive self-foaming cleansing composition contains anionic surfactants, preferably contains C<sub>8</sub>-C<sub>18</sub> acyl isethionates. These esters are prepared by reaction between alkali metal isethionate with mixed aliphatic fatty acids having from 6 to 18 carbon atoms and an iodine value of less than 20. At least 75% of the mixed fatty acids have from 12 to 18 carbon atoms and up to 20 25% have from 6 to 10 carbon atoms.

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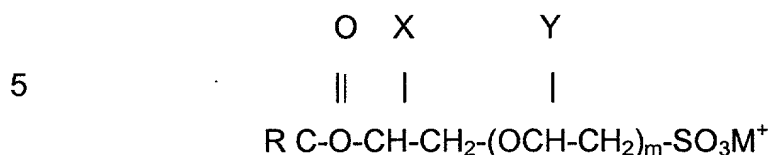
Total surfactants will generally range from about 0.5% to about 65% by weight of the self-foaming cleansing composition. Preferably, this component is present from about 2 % to about 50 % in the self-foaming 25 cleansing composition.

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The acyl isethionate may be an alkoxyated isethionate such as is described in Ilardi et al., U.S. Patent No. 5,393,466, titled "Fatty Acid Esters of

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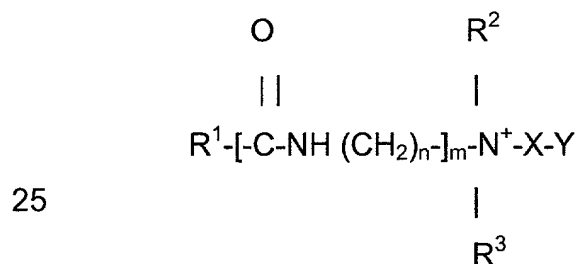
Polyalkoxylated isethonic acid; issued February 28, 1995; hereby incorporated by reference. This compound has the general formula:



wherein R is an alkyl group having 8 to 18 carbons, m is an integer from 1 to 4, X and Y are hydrogen or an alkyl group having 1 to 4 carbons and  
10 M<sup>+</sup> is a monovalent cation such as, for example, sodium, potassium or ammonium.

#### Amphoteric Surfactants

15 One or more amphoteric surfactants may be used in this invention. Such surfactants include at least one acid group. This may be a carboxylic or a sulphonic acid group. They include quaternary nitrogen and therefore are quaternary amido acids. They should generally include an alkyl or alkenyl group of 7 to 18 carbon atoms. They will usually comply with an overall structural  
20 formula:



where R<sup>1</sup> is alkyl or alkenyl of 7 to 18 carbon atoms;

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$R^2$  and  $R^3$  are each independently alkyl, hydroxyalkyl or carboxyalkyl of 1 to 3 carbon atoms;

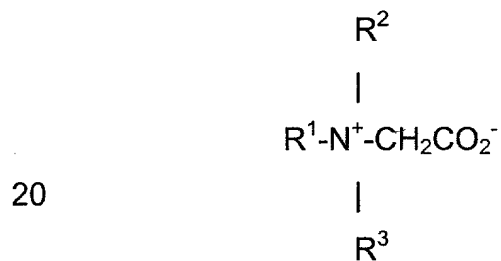
5  $n$  is 2 to 4;

$m$  is 0 to 1;

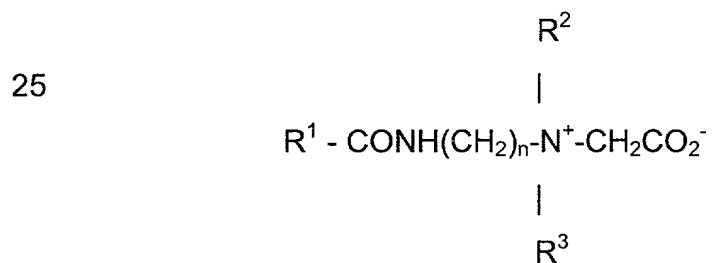
10  $X$  is alkylene of 1 to 3 carbon atoms optionally substituted with hydroxyl, and

$Y$  is  $-\text{CO}_2^-$  or  $-\text{SO}_3^-$

15 Suitable amphoteric surfactants within the above general formula include simple betaines of formula:



and amido betaines of formula:



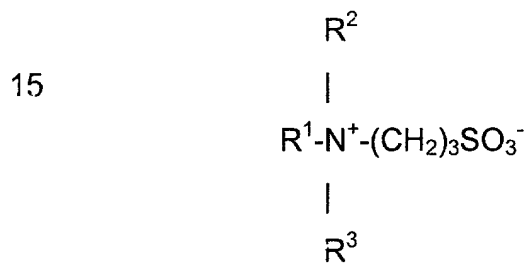


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where n is 2 or 3.

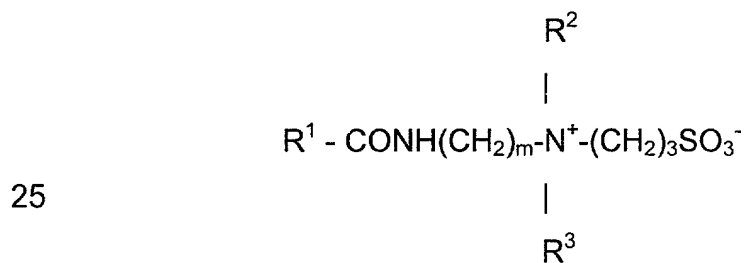
5                    In both formulae  $R^1$ ,  $R^2$  and  $R^3$  are as defined previously.  $R^1$  may in particular be a mixture of  $C_{12}$  and  $C_{14}$  alkyl groups derived from coconut oil so that at least half, preferably at least three quarters of the groups  $R^1$  have 10 to 14 carbon atoms.  $R^2$  and  $R^3$  are preferably methyl.

10                   A further possibility is that the amphoteric detergent is a sulphobetaine of formula:



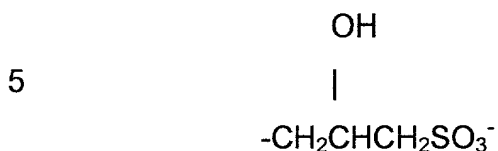
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or



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where m is 2 or 3, or variants of these in which  $-(CH_2)_3 SO_3^-$  is replaced by



10 In these formulae  $R^1$ ,  $R^2$  and  $R^3$  are as discussed previously.

Amphoacetates and diamphoacetates are also intended to be covered in possible zwitterionic and/or amphoteric compounds which may be used such as e.g., sodium lauroamphoacetate, sodium cocoamphoacetate, and blends thereof, and the like.

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### Nonionic Surfactants

20 One or more nonionic surfactants may also be used in the self-foaming cleansing composition of the present invention.

The nonionics which may be used include in particular the reaction products of compounds having a hydrophobic group and a reactive hydrogen atom, for example aliphatic alcohols, acids, amides or alkylphenols with alkylene oxides, especially ethylene oxide either alone or with propylene oxide. Specific nonionic detergent compounds are alkyl ( $C_6$ - $C_{22}$ ) phenols ethylene oxide condensates, the condensation products of aliphatic ( $C_8$ - $C_{18}$ ) primary or secondary linear or branched alcohols with ethylene oxide, and products made

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by condensation of ethylene oxide with the reaction products of propylene oxide and ethylenediamine. Other so-called nonionic detergent compounds include long chain tertiary amine oxides, long chain tertiary phosphine oxides and dialkyl sulphoxide, and the like.

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The nonionic may also be a sugar amide, such as a polysaccharide amide. Specifically, the surfactant may be one of the lactobionamides described in U.S. Patent No. 5,389,279 to Au et al. titled "Compositions Comprising Nonionic Glycolipid Surfactants issued February 14, 1995; which is hereby incorporated by reference or it may be one of the sugar amides described in Patent No. 5,009,814 to Kelkenberg, titled "Use of N-Poly Hydroxyalkyl Fatty Acid Amides as Thickening Agents for Liquid Aqueous Surfactant Systems" issued April 23, 1991; hereby incorporated into the subject application by reference.

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#### Cationic Skin Conditioning Agents

An optional component in compositions according to the invention is a cationic skin feel agent or polymer, such as for example cationic celluloses.

Cationic cellulose is available from Amerchol Corp. (Edison, NJ, USA) in their Polymer JR (trade mark) and LR (trade mark) series of polymers, as salts of hydroxyethyl cellulose reacted with trimethyl ammonium substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 10. Another type of cationic cellulose includes the polymeric quaternary ammonium salts of hydroxyethyl cellulose reacted with lauryl dimethyl ammonium-substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 24. These materials are available from Amerchol Corp. (Edison, NJ, USA) under the tradename Polymer LM-200.

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5 A particularly suitable type of cationic polysaccharide polymer that can be used is a cationic guar gum derivative, such as guar hydroxypropyltrimonium chloride (Commercially available from Rhone-Poulenc in their JAGUAR trademark series). Examples are JAGUAR C13S, which has a low degree of substitution of the cationic groups and high viscosity, JAGUAR C15, having a moderate degree of substitution and a low viscosity, JAGUAR C17 (high degree of substitution, high viscosity), JAGUAR C16, which is a hydroxypropylated cationic guar derivative containing a low level of substituent groups as well as cationic quaternary ammonium groups, and JAGUAR 162 which is a high transparency, medium viscosity guar having a low degree of substitution.

15 Particularly preferred cationic polymers are JAGUAR C13S, JAGUAR C15, JAGUAR C17 and JAGUAR C16 and JAGUAR C162, especially Jaguar C13S. Other cationic skin feel agents known in the art may be used provided that they are compatible with the inventive formulation.

#### 20 Cationic Surfactants

One or more cationic surfactants may also be used in the inventive self-foaming cleansing composition.

25 Examples of cationic detergents are the quaternary ammonium compounds such as alkyldimethylammonium halogenides.

Other suitable surfactants which may be used are described in U.S. Patent No. 3,723,325 to Parran Jr. titled "Detergent Compositions

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Containing Particle Deposition Enhancing Agents" issued March, 27, 1973; and "Surface Active Agents and Detergents" (Vol. I & II) by Schwartz, Perry & Berch, both of which are also incorporated into the subject application by reference.

- 5                    In addition, the inventive self-foaming cleansing composition composition of the invention may include 0 to 15% by wt. optional ingredients as follows:

- perfumes; sequestering agents, such as tetrasodium  
10 ethylenediaminetetraacetate (EDTA), EHDP or mixtures in an amount of 0.01 to 1%, preferably 0.01 to 0.05%; and coloring agents, opacifiers and pearlizers such as zinc stearate, magnesium stearate, TiO<sub>2</sub>, EGMS (ethylene glycol monostearate) or Lytron 621 (Styrene/Acrylate copolymer) and the like; all of which are useful in enhancing the appearance or cosmetic properties of the  
15 product.

- The compositions may further comprise antimicrobials such as 2-hydroxy-4,2', 4' trichlorodiphenylether (DP300); preservatives such as dimethyloldimethylhydantoin (Glydant XL1000), parabens, sorbic acid etc., and  
20 the like.

- The compositions may also comprise coconut acyl mono- or diethanol amides as suds boosters, and strongly ionizing salts such as sodium chloride and sodium sulfate may also be used to advantage.

- 25                    Antioxidants such as, for example, butylated hydroxytoluene (BHT) and the like may be used advantageously in amounts of about 0.01% or higher if appropriate.

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Humectants such as polyhydric alcohols, e.g. glycerine and propylene glycol, and the like; and polyols such as the polyethylene glycols listed below and the like may be used.

5	Polyox WSR-205	PEG 14M,
	Polyox WSR-N-60K	PEG 45M, or
	Polyox WSR-N-750	PEG 7M.

10           The emollient "composition" may be a single benefit agent component or it may be a mixture of two or more compounds one or all of which may have a beneficial aspect. In addition, the benefit agent itself may act as a carrier for other components one may wish to add to the self-foaming cleansing composition composition.

15           A blend of a hydrophobic and hydrophilic emollients may be used. Preferably, hydrophobic emollients are used in excess of hydrophilic emollients in the inventive self-foaming cleansing composition. Most preferably one or more hydrophobic emollients are used alone. Hydrophobic emollients are

20           preferably present in a concentration greater than about 10 % by weight, more preferably about 12 % by weight. The term "emollient" is defined as a substance which softens or improves the elasticity, appearance, and youthfulness of the skin (stratum corneum) by either increasing its water content, adding, or replacing lipids and other skin nutrients; or both, and keeps it soft by retarding

25           the decrease of its water content.

Useful emollients include the following:

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(a) silicone oils and modifications thereof such as linear and cyclic polydimethylsiloxanes; amino, alkyl, alkylaryl, and aryl silicone oils;

(b) fats and oils including natural fats and oils such as jojoba, soybean, sunflower, rice bran, avocado, almond, olive, sesame, persic, castor, coconut, mink oils; cacao fat; beef tallow, lard; hardened oils obtained by hydrogenating the aforementioned oils; and synthetic mono, di and triglycerides such as myristic acid glyceride and 2-ethylhexanoic acid glyceride;

(c) waxes such as carnauba, spermaceti, beeswax, lanolin, and derivatives thereof;

(d) hydrophobic and hydrophilic plant extracts;

(e) hydrocarbons such as liquid paraffins, vaseline, microcrystalline wax, ceresin, squalene, pristan and mineral oil;

(f) higher fatty acids such as lauric, myristic, palmitic, stearic, behenic, oleic, linoleic, linolenic, lanolic, isostearic, arachidonic and polyunsaturated fatty acids (PUFA);

(g) higher alcohols such as lauryl, cetyl, stearyl, oleyl, behenyl, cholesterol and 2-hexydecanol alcohol;

(h) esters such as cetyl octanoate, myristyl lactate, cetyl lactate, isopropyl myristate, myristyl myristate, isopropyl palmitate, isopropyl adipate, butyl stearate, decyl oleate, cholesterol isostearate, glycerol monostearate, glycerol distearate, glycerol tristearate, alkyl lactate, alkyl citrate and alkyl tartrate;

(i) essential oils and extracts thereof such as mentha, jasmine, camphor, white cedar, bitter orange peel, ryu, turpentine, cinnamon, bergamot, citrus unshiu, calamus, pine, lavender, bay, clove, hiba, eucalyptus, lemon, starflower, thyme, peppermint, rose, sage, sesame, ginger, basil, juniper, lemon grass, rosemary, rosewood, avocado, grape, grapeseed, myrrh, cucumber, watercress, calendula, elder flower, geranium, linden blossom, amaranth,

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seaweed, ginko, ginseng, carrot, guarana, tea tree, jojoba, comfrey, oatmeal, cocoa, neroli, vanilla, green tea, penny royal, aloe vera, menthol, cineole, eugenol, citral, citronelle, borneol, linalool, geraniol, evening primrose, camphor, thymol, spirantol, penene, limonene and terpenoid oils;

5 (j) lipids such as cholesterol, ceramides, sucrose esters and pseudo-ceramides as described in European Patent Specification No. 556,957;

(k) vitamins, minerals, and skin nutrients such as milk, vitamins A, E, and K; vitamin alkyl esters, including vitamin C alkyl esters; magnesium, calcium, copper, zinc and other metallic components;

10 (l) sunscreens such as octyl methoxyl cinnamate (Parsol MCX) and butyl methoxy benzoylmethane (Parsol 1789);

(m) phospholipids;

(n) antiaging compounds such as alpha hydroxy acids, beta hydroxy acids; and

15 (o) mixtures of any of the foregoing components, and the like.

Preferred emollient benefit agents are selected from triglyceride oils, mineral oils, petrolatum, and mixtures thereof. Further preferred emollients are triglycerides such as sunflower seed oil.

20

Dispensing systems; Propellants and Post-foaming agents:

Suitable aerosol barrier can dispensing systems include bag in can, pressurized bladder type packaging, piston type packaging, and the like.

25 Especially preferred is the piston type packaging for ease of production and low cost.



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Dispensing systems according to the invention have two associated gases; a propellant gas and a foaming agent. The propellant gas is that which is contained within the can, but acts against the outside of the bag, barrier, or piston in which the composition is contained to dispense the composition when  
5 an actuator on the can is used. The propellant gas can be any suitable gas, but is conveniently a liquifiable volatile hydrocarbon, such as isobutane or blends thereof, though any propellant which would function to dispense the composition would be suitable. The propellant gas is present in the packaged composition at any required and suitable level, but is typically present in the  
10 packaged product at levels sufficient to produce a satisfactory dispensing pressure, which will typically be 29-174 psi, more preferably 43-135 psi.

The other component required in the inventive compositions according to the invention is a foaming agent. The foaming agent is present in  
15 compositions according to the invention to allow the composition, which is dispensed in the form of a shear thinning lotion, once dispensed to generate a foam, as it evaporates on contact with room temperature air, even more rapidly on contact with a skin surface. The generation of a foam provides a product which has various desirable consumer attributes, including ease of  
20 handling and spreading, and desirable sensory properties.

Suitable post foaming agents for inclusion in compositions according to the invention can include any gas or volatile liquid that is soluble or dispersible in the composition. Especially preferred are hydrocarbons, such as isobutane  
25 and isopentane. Foaming agents are present in the inventive compositions according to the invention at levels of 3-20%, preferably 4-15%, and more preferably 5-10% by weight of the total composition.

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Suitable foaming agents preferably are capable of being contained in compositions according to the invention as liquids, which may have been formed under the pressure to which the packaged composition has been subjected. As such, it is also preferable that packages in which the inventive  
5 composition is stored have little to no head space, to prevent the premature evaporation of the post foaming agent.

The composition may also comprise decorative or functional particulates including speckles, coloured or reflective particles, or shaped particles,  
10 encapsulated beads, sponge, and the like.

Except in the operating and comparative examples, or where otherwise explicitly indicated, all numbers in this description indicating amounts of material ought to be understood as modified by the word "about".  
15

The following examples will more fully illustrate the embodiments of this invention. All parts, percentages and proportions referred to herein and in the appended claims are by weight unless otherwise illustrated. Physical test methods are described below:

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Example 1 - Skin Moisturization Clinical Test Results:

Skin moisturization clinical test results (see methodology below) were evaluated for inventive and comparative formulations described in tables 4 and 5. Processes for preparing the comparative isotropic gel formulations of table 5 are described in Table 7. The skin moisturization clinical test results are displayed in Tables 1 – 3 respectively; and graphically depicted according to visual evaluation, Figure 1; Corneometer data, Figure 2; and Skicon data, Figure 3,

Comparative sample A with a hydroxypropylmethyltrimonium chloride level of 0.27 wt% and hydroxymethylcellulose at 0.15% filled into the bag-in-can container was tested against Inventive examples GG, HH, II, (also filled into bag-in-cans) and II neat in both the skin moisturization clinical and sensory tests (see Example 2).

Figure 1 (Visual dryness) shows that the most effective (least visual dryness) product(s) are Inventive II Neat, Inventive Post foamer II, and Inventive Post-foamer HH. Inventive Post-foamer GG is slightly less effective (and more visual dryness), but this difference is not statistically significant. However, Comparative Post-foamer A is significantly less effective (with significantly more visual dryness).

Skicon (Fig. 2) and corneometer (Fig. 3) both are measures of skin hydration (and thus moisturization) based on measuring electrical properties using different parameters. The skicon measures skin conductance which emphasizes the hydration at or very close to the skin surface. The data for all

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samples except comparative post-foamer A are clustered together ; the data for comparative post-foamer A shows that using it leads to significantly less hydrated skin compared to the other samples.

- 5            Corneometer (Fig. 3) measures capacitance which is primarily a bulk effect which is less sensitive to surface behavior. However, the least moisturizing product (lowest trace on Fig. 3) is that of comparative post-foamer A.

10

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Example 2 - Lather And Sensory Effect:

The lather and sensory effect of the inventive composition HH in bag-in-can was compared to an isotropic gel comparative formulation A in bag-in-can using monadic test subject data from a proto-monadic test of 89 subjects with a balanced order of presentation. Each product was evaluated for 1 week. Subjects showered at least four times a week with each sample. It was seen that the inventive product provides a "dense, soft foam" based on the following data:

87% agree completely or somewhat that it has rich creamy lather  
78% agree completely or somewhat that has rich luxurious cleansing foam  
69% agree completely or somewhat that the product has a pleasing consistency  
77% agree completely or somewhat that the product rinses quickly and easily  
76% agree completely or somewhat that it doesn't leave a residue

Sensory benefits were evaluated in monadic test having 31 subjects in the inventive cell and 47 in comparative cell. The following results were found (percent agree completely and agree somewhat):

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	Sensory benefit	Comparative	Inventive
5	Does not dry your skin	65	90
	Makes your skin feel soft	78	84
	Leaves your skin feeling silky	61	78
	Moisturizing your skin	48	81
	Leaves your skin feeling smooth	74	87
	Makes your skin feel healthy	45	65

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EXAMPLE 3 - Aerosol Piston Can Performance:

The function of inventive lamellar and comparative isotropic formulations in an aerosol piston can as described in tables 4 and 5 was  
5 examined. Processing and can filling directions are listed below.

Two failure modes became apparent when working with the comparative isotropic formulations. For proper functioning, It is critical for the lotion base and foaming agent(s) to mix well and show no separation. If there  
10 is separation, then it is possible in production to see variation from can to can in foaming agent content, or have complete separation and have no foaming agent in the can, or conversely have all foaming agent (and little or no product) in the can. Alternatively, with those isotropic samples that did mix well, blow-by was observed in the piston can. Blow-by refers to an insufficient seal of  
15 the product + foaming agent mixture at the junction between the piston and the edges of the can. If such a junction is insufficient, the propellant and the product can each "blow-by" the piston, resulting either in propellant above the piston, or product below the piston. Failed comparative cans were dissected after depressurizing by perforating the can on the "propellant side" of the  
20 piston. Usually a gel emerged from the perforation demonstrating that blow-by had occurred i.e. a substantial quantity of product was under the piston in place of the propellant.

In all the above formulations, if the base and volatile foaming agents  
25 mixed well, the cans dispensed product successfully immediately after filling. However, for all the cases above that did mix, after the stresses due to shipping, product was unable to be dispensed successfully. On depressing the actuator, either no material was dispensed, or a very small quantity was

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dispensed (total foam; not post-foaming), or propellant gas came out (hissing with no product). Such a failure to dispense may imply that blowby occurred sometime during the time between initial testing and after shipping.

5           A screening test was developed wherein 6 wt% of heptane as a model for a hydrocarbon foaming agent, was added to 94 wt% of the lotion base concentrate. If the two mixed well, it was considered to be a potential candidate for filling into the piston cans. Although it was observed that our thicker systems did in fact mix well with the hydrocarbons in lab, it was  
10       observed that any isotropic system greater than 5,000 cP (measured at 0.5RPM with a Brookfield viscometer, see procedure below) did not mix well on a plant scale with hydrocarbon foaming agents. Even lower viscosity systems, designed to thicken upon addition of hydrocarbon, were not usable in the piston can although they could be dispensed in the bag-in-can dispenser.

15           Three different methodologies were evaluated for comparative isotropic gel formulations: increasing viscosity with addition of hydrocarbon foaming agent (see e.g. PCT publication no. WO0039273), decreasing or leaving the viscosity unchanged with hydrocarbon foaming agent addition (see e.g. PCT  
20       publication no. WO9703646) and the addition of low amounts of soap (see e.g. comparative examples). All were seen to fail in the piston can. Surprisingly, the inventive lamellar shear-thinning fluid in combination with the same hydrocarbon foaming agent blends in the piston system dispensed well.

25           Ability to mix with the hydrocarbons and either build or maintain a sufficient viscosity, however, is not sufficient to predict successful dispensing from the piston can. A surprising result of our work is that we found that no matter how viscous the isotropic systems + hydrocarbons were, they failed in



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the piston can. The specific shear thinning nature of the lamellar systems was found to be important to their success in the piston can.

5 The inventive lamellar systems were found not to separate on mixing with hydrocarbons, nor did they appreciably change viscosity immediately on mixing. To better characterize the lamellar systems, their rheological behavior was evaluated as described below in a controlled stress ramp (see Figures 4 – 6).

10 A representative selection of isotropic and lamellar samples had the following shear thinning index values (which is defined below):

Table 6

15	Formula	STI Value
	A	0.06
	AA	.67
	CC	.89
	EE	1.07
20	LL	0.68

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EXAMPLE 4 - Foam Density:

The foam density of the inventive post-foaming lamellar compositions EE and GG was compared to comparative post-foaming isotropic composition A and Bath and Body Works Foam Burst Moisturizing Body Wash in Lavender Flowers and the following mousses: Time Out Mango Ambrosia Whipped Shower ®Mousse (Sears), Victoria's Secret Garden Whipped Body Wash ® in Whispering Mist ®, John Frieda ® sheer blonde - blonde ambition ® hair mousse; using the method described below.

The density of the inventive lamellar post-foamers within the first 30 seconds of dispensing was found to be between 0.3-0.9 g/ml. After 5 minutes it ranged between 0.2-0.50 g/ml. We compared this to the comparative isotropic post-foaming body wash in the examples and measured the foam density to be as low as 0.1g/mL immediately on dispensing. It should be noted that the post-foamers all continued to evolve appreciably during the first 5 minutes after dispensing; the mousses either barely evolved, or evolved to a much lower extent, after the 5 minute period.

Table 8:

Sample	within 30 sec	after 5 min
Lamellar EE	0.72	0.46
Lamellar GG	0.92	0.43
Isotropic A	0.08	0.03
Bath and Body FoamBurst®	0.53	0.27
Time Out Shower ® Mousse	0.04	0.03
Victoria's Secret Garden®	0.07	0.05
John Frieda ® Hair Mousse	0.03	0.02

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#### EXAMPLE 5 - Oil Holding Capacity:

The oil holding capacity of isotropic systems versus lamellar systems were compared using the test method described below.

5

Three different surfactant bases (all clear, nonlamellar) were used: Suave® unperfumed base, Caress® Wild Blossom body wash, and Softsoap® hydrating body wash with moisture beads (commercially obtained).

- 10 For each kind of base, 7 samples of 200 g base were added into 600 mL beakers. To the various beakers, aliquots of 1,2,4,6,8,10 wt% sunflower oil were added. A propeller was placed into each beaker and each was mixed at 150 rpm for 5-10 minutes (until appeared uniformly opaque). These were placed into airtight plastic containers for observation at room temperature.

15

#### Results:

- Complete separation was observed in all Suave® samples except 1% after 1 week (clear on the bottom with a white creamy layer on top). The 1% sample had a very thin (~1mm) white creamy layer on top and was less  
20 creamy (but not quite clear) on the bottom. After 2 weeks complete separation was observed. Complete separation was observed in all Caress® samples after 3 days.

- After 2 weeks, swirls of translucent gel within the opaque gel were  
25 observed in all the Softsoap® samples. This is indicative of the beginning stages of separation.

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EXAMPLE 6 - Foam Stability :

5 The stability of comparative post-foaming isotropic body wash A packaged in a bag-in-can was compared to an inventive post-foaming lamellar composition GG. In this case the comparative sample contained no oil emollients but contained fatty acids which are added as lubricants.

10 Upon comparing the foam produced (see method below) the inventive lamellar height was 1.5 inches and the isotropic height was 3.5 inches showing decreased and more controlled foaming for the inventive composition (See figure 7). Both products contained 8% of a 75/25 isopentane/ isobutane blend as the foaming agent.

15 Once dispensed from bag-in-can type dispensers, it was also observed that in the comparative isotropic systems (such as A), there is "skunktailing" where the edges of the dispensed stream foam more readily than the center of the stream. This is an indication of improperly dispersed hydrocarbon propellant, and is less observed in the inventive lamellar systems. Foaming of the inventive composition is also delayed compared to comparative post-foamers but can be accelerated with shearing such as the action of a person  
20 lathering it on their body.

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## DESCRIPTION OF TEST METHODS:

### 1) SKIN MOISTURIZATION CLINICAL TEST METHODOLOGY

The objective of this study was to explore the conditions under which  
5 positive moisturization occurs and to compare various inventive and  
comparative formulations under those conditions. The effects of washing with  
the various skin cleansing formulations on the dry skin of the lower legs was  
measured visually and with non-invasive instrumental assessments as  
described below.

#### 10 Experimental Method

A randomized, double blind study was used. The study involved a two-  
day conditioning phase during which the subjects used a cleanser. The test  
phase of the study consisted of one or two applications of a test material on  
one or all of the sites with post-treatment visual and instrumental evaluations  
15 at 1, 2, 4, 6, 8 and 24 hours after the final application. In this procedure, each  
outer, lower leg was divided into three sites (upper, middle and lower) and the  
sites were washed with a designated cleanser. One or two of the six sites  
remained untreated controls that were included in the randomization scheme  
as a treatment. The observed effects indicated the point at which skin barrier  
20 restoration and moisturization has been achieved.

Twenty (20) subjects between 18-65 years of age in good health were  
selected who were prone to developing dry skin on their legs in the absence of  
using a moisturizer.

Subjects refrained from the following: 1) using creams, lotions,  
25 moisturizers, bath oils, additives, or any other skin products, other than those

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provided, on their lower legs for the conditioning phase and during the study; 2) consuming hot or caffeine containing beverages or smoking for one hour prior to instrumental measurements of their skin. 3) using any appliances, (wash cloth, sponge, etc.) on the test sites for duration of the study, and 5 refrained from shaving their legs 30 hours before a scheduled appointment throughout the study, 4) wetting the test sites within 3 hours of evaluation; 5) excessive sun exposure (resulting in sunburn) on the lower legs during the conditioning and treatment phases of the study.

10 All subjects have dryness scores of  $< 3.0$  and erythema scores of  $< 1.0$  on the test sites, and be free of cuts or abrasions on the outer, lower legs to be included into the conditioning phase of the study; and, have dryness scores of at least 1.0 but no greater than 2.5 with a maximum of 1.0 difference among all sites, and erythema scores of  $< 1.0$  following the conditioning phase to be included in the test phase of the study.

15 For each subject, each outer, lower leg was divided into three sites, 2.5 by 2.5 inch squares (upper, middle and lower) for a total of 6 test sites per subject. One or two of the sites was left untreated and included in the randomization of products. For studies utilizing instruments, baseline instrumental measurements were taken on each test site. The baseline 20 measurements for the sites must be within 100 units of each other for Skicon readings, within 15 units of each other for the Corneometer and Evaporimeter values to be included in the study.

The sites were treated once or twice with the designated amount of test material for 10 seconds. Cleansing products remained on the test sites for a 25 maximum of 90 seconds. Sites were rinsed for 30 seconds each, ensuring the test material from one site did not contaminate another site. After rinsing, the test sites were gently dried with a paper towel. The application consists of

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dosing with up to 5 different test materials on the designated sites, one material per test site, and one or two untreated sites. The following wash procedure was performed:

1. The test site is wet on the outer, lower leg with warm water (92°  
5 ± 4° F).
2. Fingers are wetted and the test material is dispensed to the test site.
3. Application is made by gently gliding fingers from side to side  
10 over the designated test site for the designated period of time at a rate of approximately one stroke/second (a stroke is from front to back of leg and then again to the front) ensuring not to go outside of the designated test site. Equal pressure is applied to all test sites. The sites may be washed simultaneously.
4. The test site is rinsed with warm water, patted dry with a soft  
15 disposable towel, and the procedure is repeated for the other test sites. When benefit agents or moisturizers are being tested, they will not be rinsed off the site or dried with a towel but will be allowed to air-dry for a minimum of two minutes.

#### Evaluation Methods

##### Visual Assessment

- 20 The scale as shown in Table I was used to assess the test sites for dryness, and is a 0 to 4 scale with half-point increments. Initial visual assessments were made prior to the conditioning phase to ensure that none of the subjects exceed the maximum dryness and erythema scores set for inclusion.

25

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Table 1

Grade	Dryness Scale	Erythema Scale
0.0	No dryness	No erythema
0.5	Perceptible dryness, fine white lines	
1.0	Fine dry lines, white powdery look and/or some uplifting flakes, on less than 30% of the test site	Mild erythema
1.5	More uniform flaking, covering 30-50% of the test site	
2.0	Uniform, marked flaking covering more than 50% of the test site area and/or isolated scaling	Moderate confluent Erythema
2.5	Slight to moderate scaling	
3.0	Moderate to severe scaling with some uplifting of the scales	Marked erythema
3.5	Severe scaling and/or slight fissuring	
4.0	Severe scaling and severe fissuring	Deep erythema

Baseline visual assessments were made prior to the start of the product application phase and thereafter, immediately before each of the instrumental assessments, to evaluate skin dryness and erythema. One trained evaluator

5 conducted all visual evaluations during the product application phase. The evaluator examined both lower legs with the aid of an illuminated magnifying lamp with a 3 diopter lens and a shadow-free circular cool white fluorescent light source. Half-point increments for erythema were used for responses not warranting a full point increase. To maintain the evaluator's blindness to

10 product assignment, the visual assessments were conducted in a separate area away from the product application area.



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An endpoint score was reached when a subject develops an erythema or dryness grade of 3.0 or higher on any test site. Any condition more than a 4.0 was assigned a 4.0.

#### Instrumental Assessment

- 5 All Instrumental evaluations were taken following a 30-minute acclimation period. Indoor humidity and temperature was recorded. Instrumental measurements were taken at some or all of the following time points: 0, 1, 2, 4, 6, 8 and 24 hours after product application. Instruments used for this protocol include: ServoMed Evaporimeter with EPI or EP2 probe,
- 10 Corneometer
- CM820 and the Skicon Skin Hygrometer with the MT-8C probe. The room temperature was maintained at 68° to 77° F and 30% to 40 % Relative Humidity.

#### Within Test Product Effects

- 15 The effects of each test product and the untreated site were examined by comparing the clinical grade at each time point versus the baseline clinical grade using the Wilcoxon Signed-Rank test, Pratt-Lehmann Version, as documented in Lehmann, E.L., Nonparametrics Statistical Methods Based on Ranks, pg. 123, Holden-Day, Inc., and McGraw-Hill Book Co., 1975.
- 20 Statistical significance was determined at the 90% confidence level ( $p \leq 0.10$ ).

#### Between Test Product Effects

At each evaluation point, the Pratt/Lehmann Version of the Wilcoxon Signed-Rank test was conducted on the difference in clinical grades (evaluation - baseline.) for each pair of treatments using the subject as a

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block. The untreated site was considered as a treatment in this analysis, indicating comparisons of the treatments within a group.

For the instrumental analysis data, the same comparisons were made using parametric statistical methods. The TEWL, conductance and capacitance measurements were averaged separately for each subject, site and session. For all treatments, treatment differences were statistically compared using an analysis of variance with the subject as a block. If overall statistical significant differences were detected ( $p < 0.10$ ), pairwise treatment comparisons were implemented by comparing the least square means using Fisher's Least Significant Difference.

The following references are herein included in their entirety: 1) K.D. Ertel, G.H. Keswick, and P.B. Bryant: "Forearm Controlled Application Technique for Estimating the Relative Mildness of Personal Cleansing Products", J.Soc. Cosmet Chem., 46, 67-76, 1995. 2) M.F Lukacovic, F.E. Dunlap, S.E. Michaels, M.O. Visscher, and D.D. Watson: "Forearm Wash Test to Evaluate the Mildness of Cleansing Products", J.Soc. Cosmet. Chem., 39, 355-366, 1988. 3) P.T. Sharko, R.I. Murahata, J.J. Leyden, and G.I. Grove: "Ann Wash with Instrumental Evaluation - A Sensitive Technique for Differentiating the Initiation Potential of Personal Washing Products", J. Dermalclinical Eval. Soc., 2, 19-27, 1991. 4) G.W. Snedecor. and W. G. Cochran: Statistical Methods. Ames, Iowa: The Iowa State University Press, pp. 84-86, 1980.

## 2) Aerosol can filling procedure:

The neat cleansing lotion is prepared for filling into aerosol piston cans. Manual filling consists of chilling the product and the volatile foaming

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agent(s) to 0°C, mixing the two in a beaker (e.g. 8% pure isopentane or mixtures of isopentane and isobutane or any suitable foaming agent) then adding to the can, and crimping the valve on the can. Propellant is gassed through the bottom with commercially available propellants such as A31 or

- 5 A46. The numerical designation reflects the approximate vapor pressure in psig. Such vapor pressures can be achieved for example using the following in various combinations: isobutane, n-butane, and propane, and the like. Production filling may utilize in line blending technology of the two components before they reach the filling machine.

10

For filling into bag-in-can, the cans are first pressurized (using for example an "Undercup" Crimper P 2002-500 available from Pamasol, Switzerland) with compressed air at 2.5 bar, or a liquifiable volatile hydrocarbon or other suitable propellant material and sealed by insertion of

15 the valve/diptube/laminated bag assembly before adding the lotion base-foaming agent mixture. Base product and foaming agent are filled into the bag through the valve in a single operation by use of a machine designed for such purpose, such as the KP Aerofill System from Kartridge Pak.

- 20 3) Rheological Test Methods:

A. Controlled shear stress

- Viscosity data for a stress sweep was collected using a Carri-Med CSL-100 rheometer. The experiments were run in controlled temperature, shear stress ramp between two inputted shear stress values. A cone-in-plate
- 25 geometry was used with a stainless steel 4 cm 2 degree cone.

Procedure: Power on the rheometer. Verify that the "house" air via an air filter is supplying the air bearing. Remove protective cover over the air bearing. With the set screw, attach the stainless 4 cm 2 degree cone to finger

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tight. Set the measurement temperature to 25°C and autozero the gap, after which the gap will be set to the geometry gap of 48 micrometers.

Measurements were performed at 4°C in order to characterize samples that have already been mixed with hydrocarbons that are volatile at room temperature. To accomplish this, cans of the products of interest are placed in 4°C refrigerators overnight. The measurement temperature is inputted into the rheometer, the cone is refrigerated until immediately before using, and autozeroing of the gap with the cone and plate is done at 4°C.

Lower the plate and place about 1 cc of sample on the plate using a Teflon spatula. Raise the plate to meet the cone such that there will be minimal overflow of sample around the outside of the cone. Place a cover around the sample measurement area to prevent excessive moisture loss.

Set the experimental conditions (e.g. from 0-30 Pa to be sampled in 5 minutes) and begin the experiment. Note that for samples with high yield stress (which is usually the case with the inventive lamellar samples) data may not be able to be collected at the lower shear stress values. The Rheometer will automatically begin collecting data at the lowest shear stress at which it can achieve a certain minimum angular velocity.

Viscosities at 10 and 30 Pa are used to calculate the Shear Thinning Index, or STI:

$$\text{STI} = \log [\text{viscosity at 10 Pa} / \text{viscosity at 30 Pa}].$$

#### B. Controlled shear rate

As a routine laboratory benchmark, viscosities are measured at 0.5 RPM using T-bar spindle A. Apparatus: Brookfield RVT viscometer with Helipath Accessory; chuck, weight and closer assembly for T-bar attachment; T-bar spindle A; plastic cups diameter greater than 6.35 cm (2.5 inches).

Procedure: Verify that the viscometer and the helipath stand are level by referring to the bubble levels on the back of the instrument. Connect the

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chuck/closer/weight assembly to the Viscometer (note the left-hand coupling threads). Clean Spindle A with deionized water and pat dry with a Kimwipe sheet. Slide the spindle in the closer and tighten. Set the rotational speed at 0.5 RPM. In case of a digital viscometer (DV), select the % mode and press

5 autozero with the motor switch on. Place the product in a plastic cup with inner diameter of greater than 6.35 cm (2.5 inches). The height of the product in the cup should be at least 7.6 cm (3 inches). The temperature of the product should be 25°C. Lower the spindle into the product (~6.4 mm or ¼ inches). Set the adjustable stops of the helipath stand so that the spindle

10 does not touch the bottom of the plastic cup or come out of the sample. Start the viscometer and allow the dial to make one or two revolutions before turning on the helipath stand. Note the dial reading as the helipath stand passes the middle of its downward traverse. Multiply the dial reading by a factor of 4,000 and report the viscosity reading in cps.

15

#### 4) Foam Density Determination Method

Foam density was measured using a stainless steel pycnometer. First, the pycnometer was rinsed with tap water followed by distilled water. It was

20 rinsed with acetone, dried, and allowed to come to room temperature. The empty pycnometer was tared. To calculate the volume of the pycnometer, the body of the pycnometer was carefully filled with water, avoiding air bubbles. The cover was placed on and collar screwed into place. Excess water was wiped away and the filled pycnometer was weighed.

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After cleaning and drying, the product was added to the pycnometer and the cover was carefully pushed down until seated. Excess sample expelled through the center was wiped away and the collar was screwed on.

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Excess samples was cleaned from the outside of the pycnometer and the filled pycnometer was weighed. Density is calculated using the following equation:

$$\rho = \text{g sample} / \text{g H}_2\text{O}.$$

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#### 5) Foam Comparison Procedure:

Sample one lamellar and one isotropic post-foaming composition each with 8% foaming agent (75/25 blend of isopentane and isobutane). Dispensed 10 7 g of each product into separate 7 oz cups simultaneously. After 5 minutes, the height of the foam was measured. A visual evaluation for foam quality was also done.

While this invention has been described with respect to particular 15 embodiments thereof, it is apparent that numerous other forms and modifications of the invention will be obvious to those skilled in the art. The appended claims and this invention generally should be construed to cover all such obvious forms and modifications which are within the true spirit and scope of the present invention.

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